63/1 (SEM-3) CC7/ZOOHC3076

2023

ZOOLOGY

Paper: ZOOHC3076

(Fundamentals of Biochemistry)

Full Marks: 60
Pass Marks: 24

Time: 3 hours

The figures in the margin indicate full marks for the questions

- 1. Choose the correct option of the following (any five): 1×5=5
 - (a) The biochemical test employed to identify reducing nature of sugar is
 - (i) Molich's test
 - (ii) Barfoed's test
 - (iii) Osazone test
 - (iv) Seliwanoff's test

- (b) Basicity of an amino acid depends on the presence of
 - (i) —CHO group
 - (ii) —COOH group
 - (iii) —NH group
 - (iv) —CO group
- (c) Which of the following fatty acids is an essential fatty acid?
 - (i) Stearic acid
 - (ii) Palmitic acid
 - (iii) Linoleic acid
 - (iv) Oleic acid
- (d) Turnover number of carbonic anhydrase is
 - (i) $35 \times 10^6 \text{ min}^{-1}$
 - (ii) $36 \times 10^6 \text{ min}^{-1}$
 - (iii) 35 ×10⁵ min⁻¹
 - (iv) $36 \times 10^5 \text{ min}^{-1}$

(Continued)

- (e) The ability of a sugar molecule to rotate plane polarised light into clockwise direction is called
 - (i) levorotatory
 - (ii) dextrorotatory
 - (iii) levo-dextrorotatory
 - (iv) muta-rotatory
- (f) Unsaturated fatty acid having alkyl chain on the same side is called
 - (i) cis-configuration
 - (ii) trans-configuration
 - (iii) cis-trans-configuration
 - (iv) cons-configuration
- (g) Which of the following nitrogenous base pairings is proper?
 - (i) A T
 - (ii) G C
 - (iii) T G
 - (iv) C G

- (h) The hinge region of the antibody is located in between the heavy chain
 - (i) CH₁ and CH₂
 - (ii) CH3 and CH4
 - (iii) CH₂ and CH₃
 - (iv) CH₁ and CH₃
- (i) In case of ribose-purine connection, the type of linkage is
 - (i) C₁-N₁
 - (ii) C₁-N₉
 - (iii) C₁-N₅
 - (iv) C_1-N_7
- (j) One molecule of neutral fat on hydrolysis produces
 - (i) 1 glycerol+2 fatty acids
 - (ii) 1 glycerol+3 fatty acids
 - (iii) 1 glycerol+1 fatty acid
 - (iv) 2 glycerols+3 fatty acids

- 2. Answer any *five* of the following questions (short answer-type): 2×5=10
 - (a) What are glycogenic amino acids?
 - (b) Give the absolute configuration of glyceraldehyde.
 - (c) State how pepsin is converted into an active form.
 - (d) What is phosphorylation?
 - (e) How are enzymes different from inorganic catalysts?
 - (f) What is non-coding RNA?
 - (g) Write the difference between albumins and histones.
- 3. Answer any five of the following questions: 5×5=25
 - (a) Describe lipoprotein briefly with its physiological significance.
 - (b) Describe the changes occured during denaturation of DNA helix.
 - (c) Write a brief note on the conjugated protein.
 - (d) What is antibody? Discuss its structure in brief.

- (e) Write a brief note on the mucopolysaccharide.
- (f) Describe the structure and functions of t-RNA.
- (g) Classify enzyme as per the recomendations of the International Union of Biochemistry (IUB, 1961).
- (h) Discuss the allosteric regulations of enzyme action.
- (i) Describe the effects of substrate concentration on enzyme activity.
- 4. Answer any two of the following questions: 10×2=20
 - (a) What are glycoconjugates? Describe glycolipids in brief. Write at least three biological roles of glycolipid. 1+6+3=10
 - (b) What is α-amino acid? Describe different types of amino acids based on their chemical nature.
 - (c) What is nucleotide? Describe the components of nucleotide. 2+8=10
 - (d) Derive Michaelis-Menten equation.

 State in what context the equation is useful.

 7+3=10